

# The Effect of Holding Potential on Charge Translocation by the $\text{Na}^+/\text{K}^+$ -ATPase in the Absence of Potassium

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**Abstract** The  $\text{Na}^+/\text{K}^+$ -ATPase exports  $3\text{Na}^+$  and imports  $2\text{K}^+$  at the expense of the hydrolysis of 1 ATP. In the absence of  $\text{K}^+$ , it carries on electroneutral,  $\text{Na}^+$ -dependent transient charge movement (also known as “electroneutral  $\text{Na}^+/\text{Na}^+$  exchange mode”) and produces a transient current containing faster and slower components in response to a sudden voltage step. Components with different speeds represent sequential release of  $\text{Na}^+$  ions from three binding sites. The effect of holding potential on slow charge movement was studied in the presence of different concentrations of  $\text{ADP}_i$ ,  $\text{Na}_i^+$  and  $\text{Na}_o^+$  with the intention of improving our understanding of  $\text{Na}_i^+$  binding. However, the manipulation of  $[\text{ADP}]_i$  and  $[\text{Na}^+]_i$  did not cause as pronounced changes as predicted in the magnitude of charge movement ( $Q_{\text{tot}}$ ), which indicated that our experimental conditions were not able to backwardly drive reaction across the energy barrier to  $\text{Na}_i^+$  release/rebinding steps. On the contrary, lowering  $[\text{Na}^+]_o$  caused evident dependence of  $Q_{\text{tot}}$  on holding potential, with characteristics suggesting that pumps were escaping from E2P through the uncoupled  $\text{Na}^+$  efflux activity.

**Keywords** Electrophysiology · Sodium pump · Voltage-clamp study · Biophysics

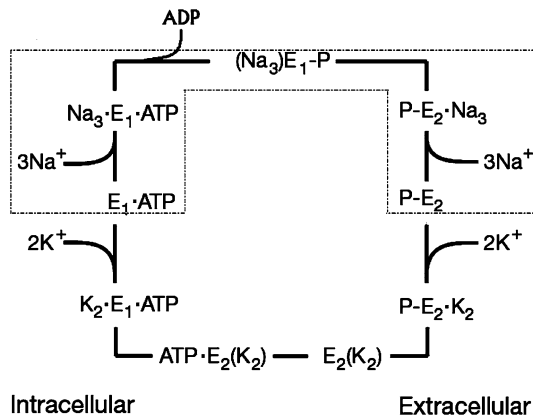
## Introduction

The  $\text{Na}^+/\text{K}^+$ -ATPase (NKA), a member of the P-type ATPase protein family, is an integral membrane protein

that exports  $3\text{Na}^+$  and imports  $2\text{K}^+$  at the expense of the hydrolysis of 1 ATP. The reaction catalyzed by NKA requires four substrates, intracellular  $\text{Na}^+$  ( $\text{Na}_i^+$ ), extracellular  $\text{K}^+$  ( $\text{K}_o^+$ ), ATP and  $\text{H}_2\text{O}$ , and produces four products, extracellular  $\text{Na}^+$  ( $\text{Na}_o^+$ ), intracellular  $\text{K}^+$  ( $\text{K}_i^+$ ), ADP and orthophosphate. NKA pumps ions in and out of the cell following a transport cycle usually referred to as the “Albers-Post scheme” (Fig. 1) (Albers 1967; Post et al. 1972; Skou and Esmann 1992). In this model, NKA can adopt two main conformations: E1 and E2. In E1, ion-binding sites face the cytoplasm and the apparent affinity for  $\text{Na}^+$  is higher than that for  $\text{K}^+$ . In E2, ion-binding sites are exposed to the extracellular solution and the apparent affinity for  $\text{K}^+$  is higher than that for  $\text{Na}^+$ . Phosphorylation of E1 by ATP promotes the occlusion (or trapping) of three intracellular  $\text{Na}^+$ , which are released to the external medium by a deocclusion transition associated with a conformational change to E2. Once two external  $\text{K}^+$  bind to two of the three ion-binding sites, they elicit dephosphorylation and  $\text{K}^+$  occlusion. ATP binding favors transition back to E1, prompting  $\text{K}^+$  release to the cytoplasm and binding of three  $\text{Na}^+$  ions to complete the cycle.

Under physiological conditions, NKA produces a net movement of one positive charge in the outward direction in each cycle. Consequently, the velocity at which the pump cycles (or turnover rate) must be sensitive to membrane potential ( $V_m$ ). Most of the voltage dependence under normal ionic conditions comes from the transport of  $\text{Na}^+$  (Rakowski 1993; Rakowski and Paxson 1988; Nakao and Gadsby 1986; Sagar and Rakowski 1994). Also, the major voltage-sensitive step occurs during  $\text{Na}_o^+$  deocclusion/occlusion and release/binding steps (Rakowski et al. 1989, 1997). This can be explained by postulating the existence of a narrow and deep access channel or “ion well” between the  $\text{Na}^+$ -binding sites within the protein and the extracellular

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**Fig. 1** Albers-Post scheme for the Na<sup>+</sup>/K<sup>+</sup>-ATPase

aqueous solution. Because the electric field is expected to drop along an access channel, negative potentials will favor binding of external Na<sup>+</sup> and, consequently, the forward pump turnover rate will be slowed down. The access channel model predicts that changes in ion concentration and membrane voltage are equivalent.

Removing K<sup>+</sup> from the experimental solutions restricts the pump to operating among the states encircled by dashed lines in Fig. 1. In this mode, NKA exchanges Na<sup>+</sup> with extracellular or intracellular medium without net movement of ions across the membrane or net hydrolysis of ATP.

At any given potential, the state occupancy of pumps is determined by the values of the rate constants. A sudden voltage step will instantly change any voltage-dependent rate constant, and the system will relax to the newly imposed steady state following a time course established by the new values of the rate constants (Nakao and Gadsby 1986). Therefore, depolarizing and hyperpolarizing steps elicited outward and inward transient currents, respectively. These currents decayed exponentially with time constants that were faster at hyperpolarizing voltages and reached a minimum at positive potentials. The voltage dependence of the charge ( $Q$ ) follows a sigmoid shape, with saturations at large hyperpolarizing and depolarizing potential. These data could be well fitted by a Boltzmann distribution, as shown in Eq. 1 (see the Appendix for more explanation),

$$\frac{Q(V) - Q_{\min}}{Q_{\text{tot}} - Q_{\min}} = \frac{1}{1 + \exp\left[\frac{zF(V_{\text{mid}} - V)}{RT}\right]} \quad (1)$$

where  $Q_{\text{tot}}$  is the total amount of charge,  $Q_{\min}$  is the minimum value of charge reached at extreme negative voltages,  $V_{\text{mid}}$  is the mid-point voltage,  $F$  is Faraday's constant,  $R$  is the gas constant,  $T$  is temperature and  $z$  is apparent valence.  $z$  can be expressed as  $n\lambda$ , where  $n$  is the Hill coefficient representing the apparent molecularity of the charge moving process and  $\lambda$  is the dielectric coefficient representing the electric field dropped between external

medium and the binding site.  $RT/zF$  represents the steepness of the curve.

With improvements of electrophysiological techniques in later studies, transient currents with fast relaxation rates were resolved. The exponential decay of the whole transient current was separated into three components, reflecting the deocclusion/release of three Na<sup>+</sup>, which occurred in sequence with increasing rate constants from the first (1,000 s<sup>-1</sup>), the second (about 6,000–20,000 s<sup>-1</sup>) to the third (10<sup>6</sup> s<sup>-1</sup>) Na<sup>+</sup> ions (Holmgren et al. 2000). The release or rebinding of the first Na<sup>+</sup> from E2P (Na3) or to E2P (Na2) forms the slow component and is the dominant charge-carrying step with a dielectric coefficient of 0.65–0.7 (Holmgren and Rakowski 2006; Holmgren et al. 2000; Wuddel and Apell 1995).

There are three other electrogenic partial reactions in the Post-Albers cycle: (1) the third intracellular Na<sup>+</sup> binding, Na<sub>2</sub> · E<sub>1</sub> · ATP + Na<sub>i</sub><sup>+</sup> → Na<sub>3</sub> · E<sub>1</sub> · ATP (Heyse et al. 1994; Wuddel and Apell 1995); (2) the conformational change, E<sub>1</sub>P(Na<sub>3</sub>) → E<sub>2</sub>P(Na<sub>3</sub>) (Heyse et al. 1994; Wuddel and Apell 1995); and (3) the extracellular K<sup>+</sup> binding, E<sub>2</sub>P + 2K<sup>+</sup> → E<sub>2</sub> · K<sub>2</sub> (Heyse et al. 1994; Sagar and Rakowski 1994; Wuddel and Apell 1995). Binding of the first two intracellular Na<sup>+</sup> ions was found to be electro-neutral since it is accompanied by exchange for bound H<sup>+</sup> (Domaszewicz and Apell 1999). According to the conservation principle, the sum of all dielectric coefficients from each electrogenic partial reaction involved in transferring each net charge across the membrane has to be 1 (Apell 2003). The values presently determined for the dielectric coefficients are the third Na<sup>+</sup> binding to the intracellular binding site, 0.25 (Domaszewicz and Apell 1999); the conformational change, 0.1–0.2 (Heyse et al. 1994); the release of the first Na<sup>+</sup> to the extracellular medium, 0.65 (Hilgemann 1994; Holmgren et al. 2000); the release of the other two Na<sup>+</sup> to the extracellular medium, each 0.1 (Heyse et al. 1994; Hilgemann 1994); and the binding of two K<sup>+</sup> to their binding sites, each –0.1 to –0.2 (Heyse et al. 1994; Rakowski et al. 1991). The sum of these values is approximately 1, which obeys the conservation principle because the pump moves one net charge per cycle (Apell 2003).

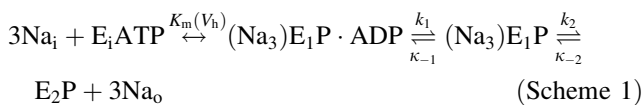
The kinetics of the internal Na<sup>+</sup> binding and release step is the least-studied part of the transport cycle. This step has a rate that is not readily detected by current techniques, so it is impossible to study its kinetics directly. However, when this step is included in the reaction scheme, it can affect the availability of pumps for extracellular Na<sup>+</sup> release. We are trying to observe it indirectly through recording its effect on extracellular Na<sup>+</sup> release.

The slow charge translocation ( $Q_{\text{slow}}$ ) that we measure comes from transitions involving extracellular Na<sup>+</sup> binding/unbinding and occlusion/deocclusion, which we could isolate fairly well by specifying the composition of the

internal and external solutions. If all pumps are confined to these states, we would expect no difference in the total amount of charge at different holding potential ( $V_h$ ). We hypothesize that by manipulating  $[\text{ADP}]_i$ ,  $[\text{Na}^+]_i$  or  $[\text{Na}^+]_o$ , we will allow pumps to exit these states through a  $V_h$ -dependent route. After holding the membrane at a potential for enough time, the state occupancy is at an equilibrium determined by the rates of all participating steps, including internal  $\text{Na}^+$ -binding/release steps that are separated from external  $\text{Na}^+$  release/rebinding by a very slow conformational change. Thus, by varying  $V_h$  and studying the corresponding change in  $Q_{\text{slow}}$ , we were expecting to infer the kinetics of internal  $\text{Na}^+$  release/rebinding. If there is another  $V_h$ -dependent route, we would get to know the nature of the voltage dependence of that route.

At a constant  $V_h$ , changes in  $[\text{Na}^+]_o$  and  $[\text{ADP}]_i$  in previous studies did not affect the magnitude of the total mobile charge (Holmgren and Rakowski 2006), but  $V_{\text{mid}}$  was shifted toward more positive voltage by increasing  $[\text{Na}^+]_o$  or  $[\text{ADP}]_i$ . A change of  $[\text{ADP}]_i$  from 0 to 4.3 mM shifted  $V_{\text{mid}}$   $16 \pm 6$  mV (Peluffo 2004), and every doubling of  $[\text{Na}^+]_o$  similarly shifted  $V_{\text{mid}}$  by  $\sim 20$  to 25 mV (Holmgren and Rakowski 2006).

At a constant  $V_h$ , changes in  $[\text{Na}^+]_i$  in previous studies did change the total amount of charge following a simple dose-response curve with an apparent affinity of 3.2 mM (Holmgren and Rakowski 2006). Because these data involve intracellular  $\text{Na}^+$  binding and occlusion, the authors based their results on the following scheme:



Therefore, charge ( $Q$ ) becomes a function of voltage and  $[\text{Na}^+]_i$  by

$$Q(V) = \left[ \frac{Q_{\text{tot}}}{1 + \exp\left(\frac{zF(V_{\text{mid}} - V)}{RT}\right)} + Q_{\text{min}} \right] \left[ \frac{[\text{Na}]_i}{[\text{Na}]_i + K_m(V_h)} \right] \quad (2)$$

where  $K_m(V_h)$  is a voltage-dependent dissociation coefficient for binding of the third intracellular  $\text{Na}^+$ . Therefore, estimating  $Q_{\text{tot}}$  at different  $V_h$  but constant  $[\text{Na}^+]_i$ , we expected to learn about the intrinsic voltage dependence of intracellular  $\text{Na}^+$  binding.

## Materials and Methods

### Oocyte Preparation and Maintenance

The charge movement ( $Q$ ) mediated by  $\text{Na}^+$  release and rebinding was measured using the cut-open oocyte

voltage-clamp technique in *Xenopus laevis* oocytes expressing *Bufo marinus* NKA  $\alpha_1$ - and  $\beta_1$ -subunits. Oocytes at stages V and VI were collected and treated with 0.3% collagenase in  $\text{Ca}^{2+}$ -free ND96 solution (96 mM NaCl, 2 mM KCl, 1 mM  $\text{MgCl}_2$ , 2.5 mM Na pyruvate, 5 mM Tris HEPES [pH 7.3], osmolarity =  $200 \pm 5$ ) for 40–60 min to remove the follicular layer. Oocytes remained viable for up to 6 days at  $15^\circ\text{C}$  in ND96 solution ( $\text{Ca}^{2+}$ -free ND96 plus 1.8 mM  $\text{CaCl}_2$ ).

### RNA Preparation and Injection

Wild-type *B. marinus* NKA  $\alpha_1$  and  $\beta_1$  cRNAs were transcribed from their corresponding cDNAs. *B. marinus* NKA  $\alpha_1$  and  $\beta_1$  cDNAs in plasmid pSD5 were transformed into competent bacteria DH5 $\alpha$ , single colonies were grown overnight and plasmids were prepared using the QIAprep Spin Miniprep Kit 27104 (Qiagen, Valencia, CA).

The extracted  $\alpha_1$  and  $\beta_1$  cDNAs were linearized by restriction enzymes *SpeI* and *BglIII* (New England Biolabs, Ipswich, MA), respectively. Linearized cDNAs were treated with proteinase K and extracted using phenol/chloroform. cRNAs were transcribed from linearized cDNAs using the MAXIScript<sup>®</sup> SP6 Kit (AM1308; Ambion, Austin, TX). Ten nanograms of  $\alpha_1$  cRNA and 1 ng of  $\beta_1$  cRNA were coinjected into oocytes 1 day after oocytes were collected. Injected oocytes were incubated in ND96 solution for 3 days to obtain good membrane expression. Oocytes were incubated overnight in 0.2  $\mu\text{M}$  ouabain containing ND96 solution to inhibit the endogenous *X. laevis* NKA current (Horisberger and Kharoubi-Hess 2002). *B. marinus*  $\alpha_1$  is more ouabain-resistant than *X. laevis*  $\alpha_1$  so that pretreatment in 0.2  $\mu\text{M}$  ouabain completely inhibits the endogenous pump current but has little effect on the introduced *Bufo* pump current, while the 1 mM ouabain applied during electrophysiological measurement inhibits both kinds of current.

### Cut-Open Setup and Recording Protocols

Saponin-permeabilized *Xenopus* oocytes mounted with guarded seals between the intracellular and extracellular compartments were used to study transient currents mediated by *Bufo* NKA (Tagliatalata et al. 1992). Compared to a regular two-microelectrode voltage-clamp recording of whole-oocyte currents, this setup has several advantages, including (1) high-frequency response and low noise recording, (2) access to the cell interior and (3) stable recording conditions lasting for several hours.

The oocyte was mounted with its animal (dark) pole oriented toward the bottom, where it was permeabilized by adding 0.5% saponin for approximately 1 min. Permeabilization resulted in a decrease in access resistance and a large increase in linear nonpump capacitance, which resulted in an

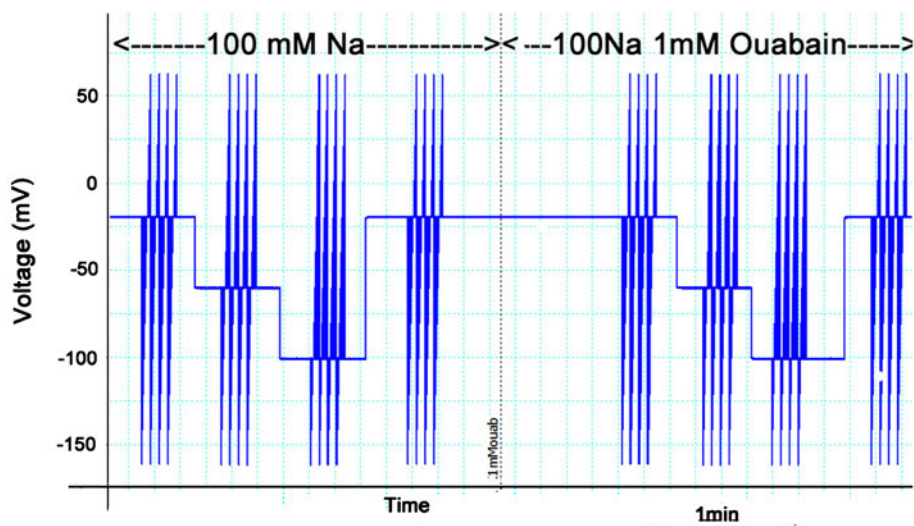
increase in the time constant of total transient currents induced by small-amplitude (20 mV) test voltage-clamp pulses (stepping from 0 to  $-20$  mV). Then, the saponin solution was replaced by the experimental intracellular solution and allowed to equilibrate until there was no change in transient current magnitude. Oocytes were stable with small background current for more than 2 h. The voltage-clamp system was obtained from Dagan (Minneapolis, MN; model CA-1 High Performance Oocyte Clamp).

The effects of three different  $V_h$  ( $-20$ ,  $-60$  and  $-100$  mV) were studied in permeabilized oocytes as  $[\text{Na}^+]_i$ ,  $[\text{ADP}]_i$  and  $[\text{Na}^+]_o$  were varied. Voltage pulses of 40 ms duration were made from  $V_h$  to command potentials over the range  $-160$  to  $+60$  mV in increments of 20 mV. Pulses were applied every 300 ms, and the current records were obtained by averaging four repetitions of the pulse protocol. Three  $V_h$  were applied in the order of  $-60$ ,  $-100$  and  $-20$  mV. To study the effect of  $V_h$  in low  $[\text{Na}^+]_o$  in detail, six  $V_h$  were applied in the order of  $-40$ ,  $-80$ ,  $0$ ,  $-60$ ,  $-100$  and  $-20$  mV. The time elapsed before and after the addition of ouabain at each  $V_h$  was the same. Two  $V_h$  of  $-20$  mV were applied to bracket the entire series of  $V_h$  both before and after the addition of ouabain (Fig. 2). The current differences between the bracketing  $V_h$  of  $-20$  mV were checked to make sure that only small time-dependent drifting happened.

Data were acquired using an analog-to-digital converter system and software (TL-1 DMA interface, 100 kHz, PCLAMP version 9; Molecular Devices, Sunnyvale, CA) running on a Dell-compatible computer system (Austin, TX). The analog signal was filtered at 5 kHz before being digitized and sampled every 20  $\mu\text{s}$ .

## Solutions

The 100 Na0 K external solution had the following composition: 100 mM Na glutamate, 20 mM tetraethylammonium (TEA) glutamate, 3 mM Mg (glutamate)<sub>2</sub>, 5 mM Ba(NO<sub>3</sub>)<sub>2</sub>, 2 mM Ni(NO<sub>3</sub>)<sub>2</sub>, 0.01 mM Gd(NO<sub>3</sub>)<sub>3</sub>, 0.3 mM niflumic acid and 10 mM Tris HEPES (pH 7.6). 25 Na0 K external solution was obtained by equimolar substitution of tetramethylammonium (TMA) for Na<sup>+</sup>. The composition of 50 Na0 K internal solution was 50 mM Na glutamate, 20 mM TEA glutamate, 10 mM MgSO<sub>4</sub>, 5 mM MgATP, 5 mM Tris ADP, 5 mM 1,2-bis(2-aminophenoxy)ethane-*N,N,N',N'*-tetraacetic acid (BAPTA), 30 mM *N*-methyl *D*-glucamine (NMDG) glutamate, 10 mM 3-(*N*-morpholino)propanesulfonic acid (MOPS, pH 7.3). Na<sup>+</sup>-free internal solution was prepared by equimolar substitution of NMDG for Na<sup>+</sup>. Intermediate intracellular Na<sup>+</sup> concentrations were obtained by mixing internal solutions containing 50 and 0 mM Na<sub>i</sub><sup>+</sup>. Note that both ATP and ADP were present in the intracellular solution, to promote electroneutral Na<sup>+</sup>/Na<sup>+</sup> exchange (De Weer 1970; Glynn and Hoffman 1971). One millimolar P1,P5-di(adenosine-5')pentaphosphate (AP5A) was included in all ADP-containing internal solutions, to prevent phosphate exchange between ATP and ADP. To study the ADP sensitivity of  $Q_{\text{slow}}$ , 5 mM phosphocreatine substituted 5 mM ADP to make sure no ADP was formed from ATP hydrolysis. The solutions were designed to minimize nonpump-mediated current. Extracellular TEA<sup>+</sup> and Ba<sup>2+</sup> were present to block passive K<sup>+</sup> conductance (Holmgren and Rakowski 1994). NMDG was used as an internal Na<sup>+</sup> substitute



**Fig. 2** Chart recording of holding potentials and voltage steps applied during a typical experiment. Oocyte was holding at  $-20$ ,  $-60$ ,  $-100$ ,  $-20$  mV in 100 or 25 mM Na<sub>o</sub><sup>+</sup> solution first. After equilibrating at each  $V_h$  for  $\sim 10$  s, four runs of voltage steps ranging from  $-160$  to  $60$  mV with increments of 20 mV were applied. Then,

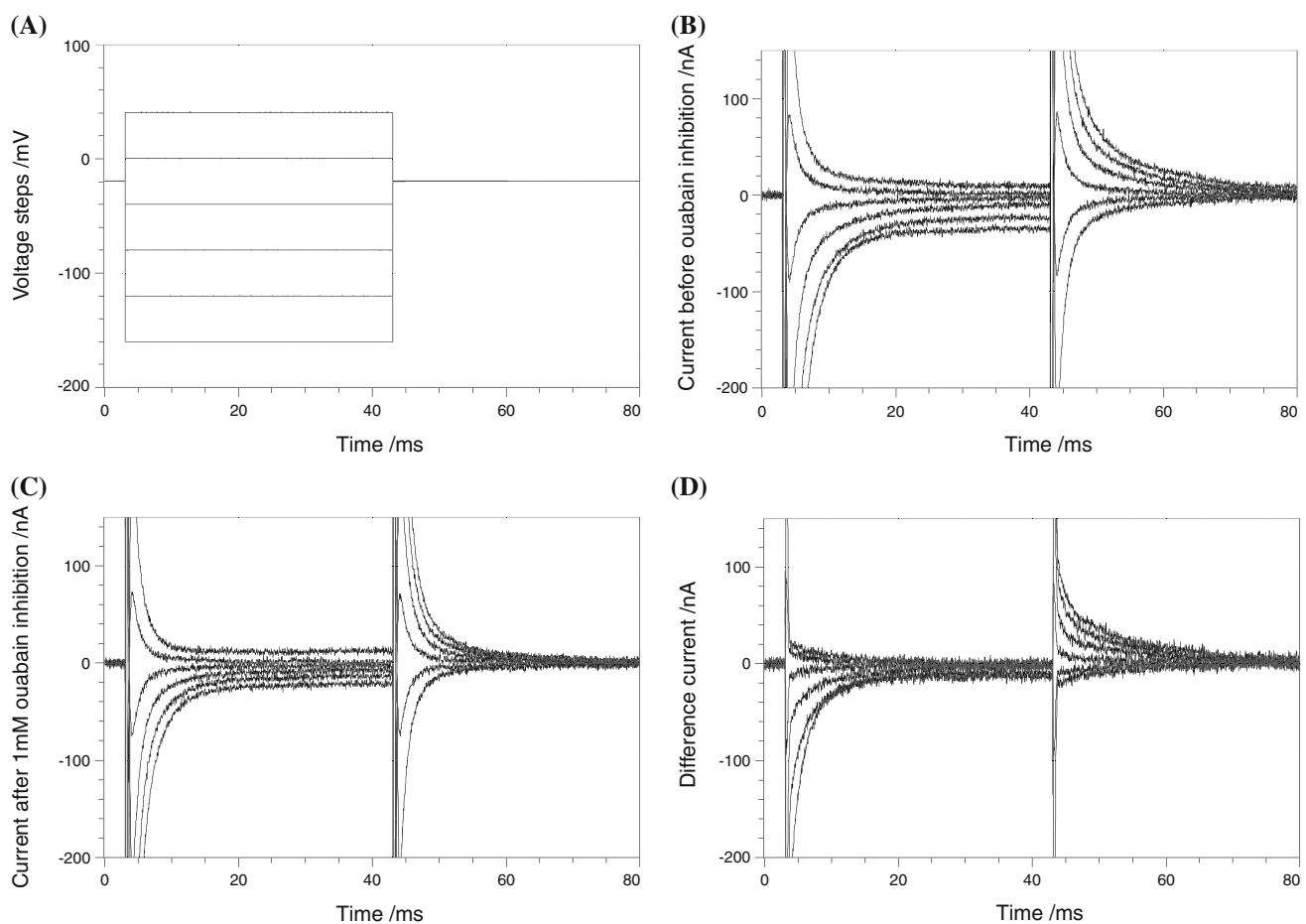
1 mM ouabain was added to the external solution. After pumps were fully inhibited, the exact same sequence of voltage changes was applied. The time elapsed from before to after the addition of ouabain at each  $V_h$  was the same

because it would be less likely to compete for  $\text{Na}^+$ -binding sites than smaller cations. The solutions were nominally chloride-free and  $\text{Ca}^{2+}$ -free, to prevent activation of  $\text{Ca}^{2+}$ -dependent anionic current. Niflumic acid was present to block  $\text{Cl}^-$  channels.  $\text{Ni}^{2+}$  and  $\text{Gd}^{3+}$  were present to block  $\text{Na}^+/\text{Ca}^{2+}$  exchange (Kimura et al. 1987) and stretch-activated cation channels (Yang and Sachs 1989), respectively.

### Transient Current Analysis

$\text{Na}^+$ -dependent transient current was calculated by subtraction of the current records obtained after arresting  $\text{Na}^+$ -dependent charge movement by the addition of 1 mM ouabain from records acquired just prior to the addition of ouabain (Fig. 3). Previous studies have demonstrated that the slow ouabain-sensitive transient current is preceded by at least one fast component with a time course close to that of the voltage step (relaxation rate  $>2,000 \text{ s}^{-1}$ ) (Hilgemann

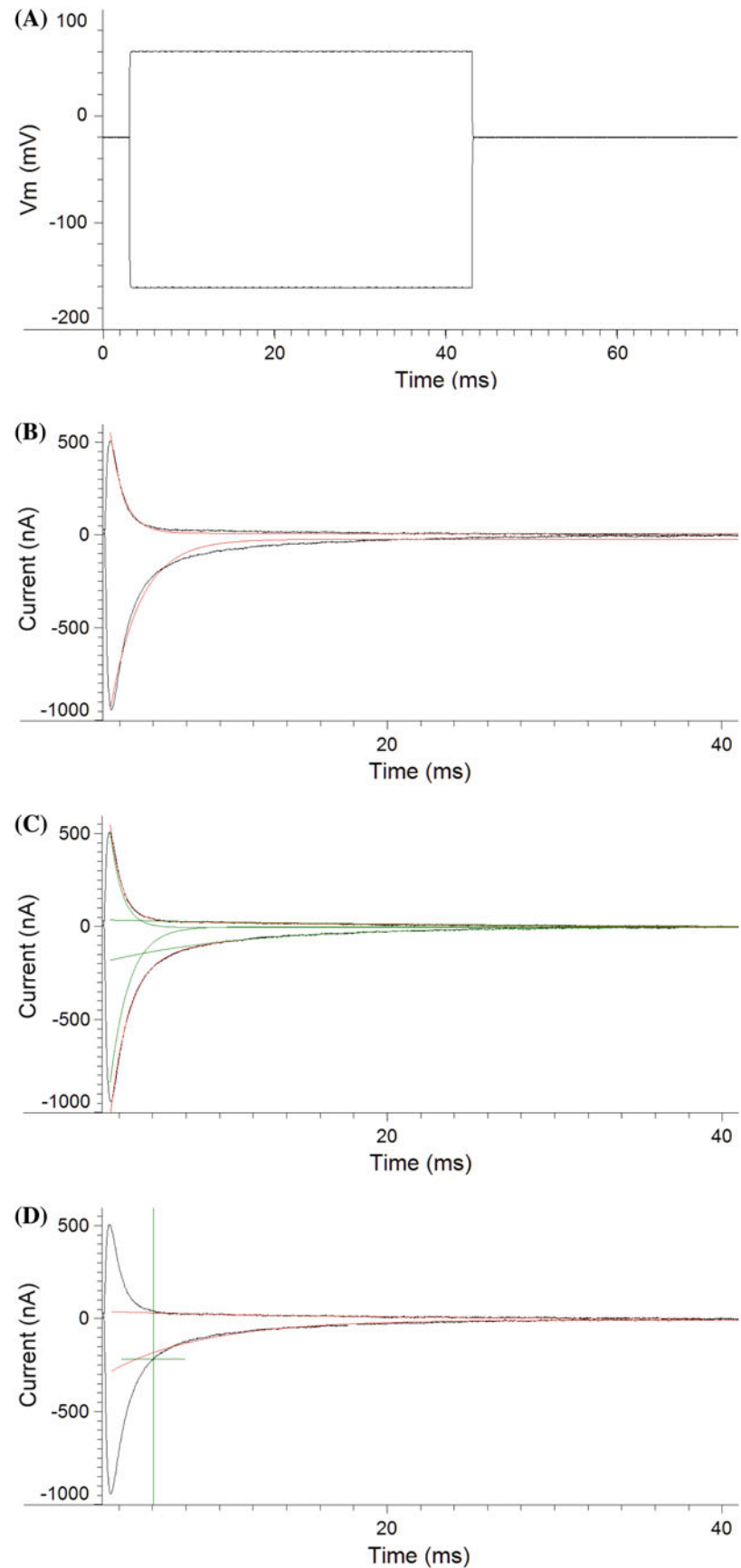
1994; Holmgren and Rakowski 2006; Holmgren et al. 2000). Therefore, to obtain the relaxation rate of slow components, transient currents were fitted with a single exponential function starting 2–4 ms after the initiation of voltage steps over a 30-ms period and the fit was extended to about 0.5 ms after the start of voltage steps to include the complete slow component without contamination by faster components. In another study of  $\text{Na}^+$ -dependent transient current, the fit was started about 0.5 ms after the beginning of the pulse (Colina et al. 2007); but in our study with *Bufo* NKA expressed in *Xenopus*, to get a good fit, the single exponential fit had to be started 2–4 ms after the beginning of the pulse (Fig. 4b, d). If the fit was started 0.5 ms after the beginning of the pulses, an exponential function with two components was needed for a good fit (Fig. 4c). This might be due to the difference in the NKAs between species. In order to observe the  $\text{Na}_o^+$ -dependent slow charge movement in full scale, the fit in our study was then extended to the beginning of the pulse (Fig. 4d).

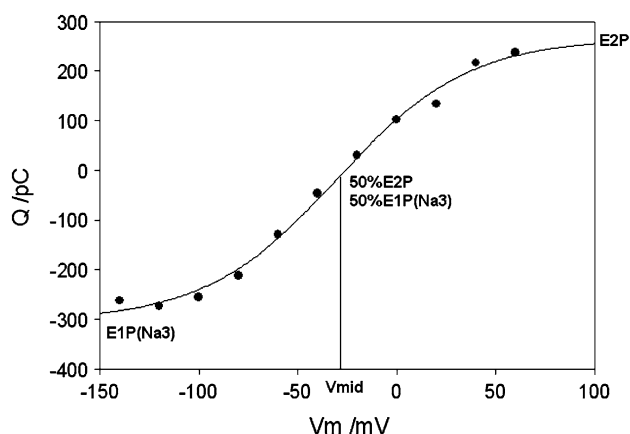


**Fig. 3** Ouabain-sensitive transient currents from a permeabilized oocyte. **a** Voltage pulse protocol. Voltage pulses were done in increments of 20 mV from  $-160$  to  $+60$  mV with duration 40 ms. For clarity, voltage pulses are shown every 40 mV. Holding potential

was  $-20$ . **b** Current time course in 100 Na0 K external solution. **c** Currents in 1 mM ouabain 100 Na0 K external solution. **d** Ouabain-sensitive currents determined by subtraction of currents in the presence of 1 mM ouabain (**c**) from those in its absence (**b**)

**Fig. 4** Different fitting methods applied to the subtracted currents obtained from an oocyte in 25 mM  $\text{Na}_i^+$  and 10 mM  $\text{Na}_o^+$ . **a** Voltage steps used to elicit currents. For simplicity, two voltages, 60 and  $-160$  mV, are shown. The 60-mV step elicited an outward transient current and  $-160$  mV elicited an inward transient current, shown in (**b–d**), where the same currents (*dark lines*) were fitted differently. **b** The subtracted currents were fitted starting 0.5 ms after the beginning of pulses with a single-exponential function. *Red lines* (*light lines* in gray scale) represent the best fit. **c** The subtracted currents were fitted starting 0.5 ms after the beginning of pulses with a two-component exponential function. *Red lines* (overlapping raw data) represent the best fit. *Green lines* (*light lines* in gray scale) represent fitted components. **d** The subtracted currents were fitted starting 3 ms after the beginning of pulses with a single-exponential function. Meanwhile, fit was extended to the beginning of pulses. *Vertical line* represents the time point when fit started. *Red lines* (*light lines* in gray scale) represent the best fit (Color figure online)





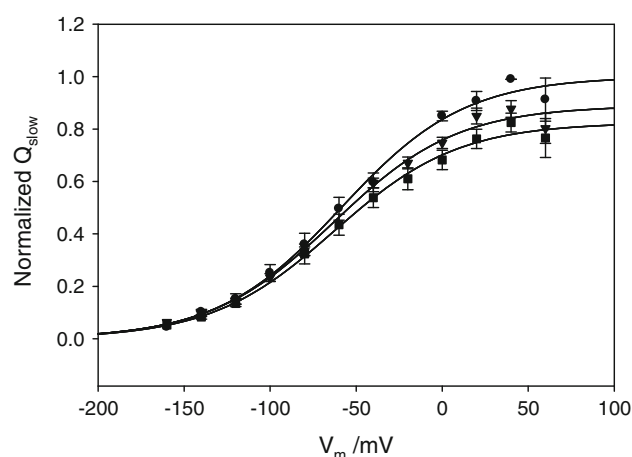
**Fig. 5** An example of voltage ( $V_m$ ) dependence of charge distribution ( $Q$ ) ( $Q$ - $V$  curve). *Solid line* represents the best-fit line of the data points using Eq. 1.  $Q_{\max}$  at extreme positive  $V_m$  represents when 100% of the measured pumps are in state E2P (no  $\text{Na}^+$  bound), and  $Q_{\min}$  at extreme negative  $V_m$  represents when 100% are in state E1P( $\text{Na}_3$ ) ( $3\text{Na}^+$  bound).  $V_{\text{mid}}$  represents when half of the pumps are in state E2P and half in E1P( $\text{Na}_3$ )

$Q_{\text{slow}}$  was then determined by multiplying the two parameters  $A$  (the fitted initial value of transient currents) and  $\tau$  (time constant of exponential decay of transient currents), obtained by the single exponential fit. All of this analysis was done with Clampfit 9.2 software. Further analysis, least-squares curve fitting and preparation of figures were done with SigmaPlot 8.0 (Systat Software, Richmond, CA). Curve-fit parameters were obtained from the least-squares fitting procedure. Experiments were performed at room temperature ( $\sim 22^\circ\text{C}$ ).

The voltage dependence of the charge movement ( $Q$ - $V$  curve) is given by Eq. 1, which has a sigmoid shape. To simplify the meaning of the curve,  $Q_{\max}$  represents the maximum occupation of pumps at  $\text{Na}^+$ -free form E2P and  $Q_{\min}$  represents the maximum occupation of pumps at  $\text{Na}^+$ -binding form E1P( $\text{Na}_3$ ) (Fig. 5).

## Results

Equation 1 was used to fit  $Q$ - $V$  curves obtained under different conditions. The value of  $V_{\text{mid}}$ , representing the position of the curve, depends on the ratio between the occupancies of E1P( $\text{Na}_3$ ) and E2P states at a certain voltage.  $V_{\text{mid}}$  is always the voltage where the ratio is 1 (Fig. 3). The  $z$  value is the apparent valence of the charge moved between those states, which floats at  $V_h$  of  $-20$  mV and then is constrained to the value obtained at  $V_h$  of  $-20$  mV for all  $Q$ - $V$  curves under the same intracellular and extracellular conditions.



**Fig. 6** The  $V_h$  dependence of the  $Q$ - $V$  curve of ouabain-sensitive transient currents in oocytes in internal solution 50 mM  $\text{Na}^+$ , 5 mM ATP, 0 ADP and 0  $\text{K}^+$  and externally perfused with 100 mM  $\text{Na}^+$  0  $\text{K}^+$ . Dots are data points (mean  $\pm$  SE) at  $V_h$  of  $-20$  mV (circles),  $-60$  mV (triangles) and  $-100$  mV (squares). *Solid lines* represent best fit to Eq. 1 with  $z$  constrained to 0.71. The  $z$  value was obtained by initially fitting the  $Q$ - $V$  curve at  $V_h$   $-20$  mV to Eq. 1 with  $z$  floating. The best-fit parameter values are  $Q_{\text{tot}} = 1.00 \pm 0.03$ ,  $V_{\text{mid}} = -58.7 \pm 5.3$  mV for  $V_h = -20$  mV;  $Q_{\text{tot}} = 0.89 \pm 0.02$ ,  $V_{\text{mid}} = -63.1 \pm 4.8$  mV for  $V_h = -60$  mV;  $Q_{\text{tot}} = 0.82 \pm 0.03$ ,  $V_{\text{mid}} = -62.6 \pm 6.4$  mV for  $V_h = -100$  mV. Data were from five oocytes of at least two different frogs

Effect of  $V_h$  on the Magnitude of  $Q_{\text{slow}}$  in 100 mM  $\text{Na}_o^+$  and 50 mM  $\text{Na}_i^+$  with or without 5 mM ADP

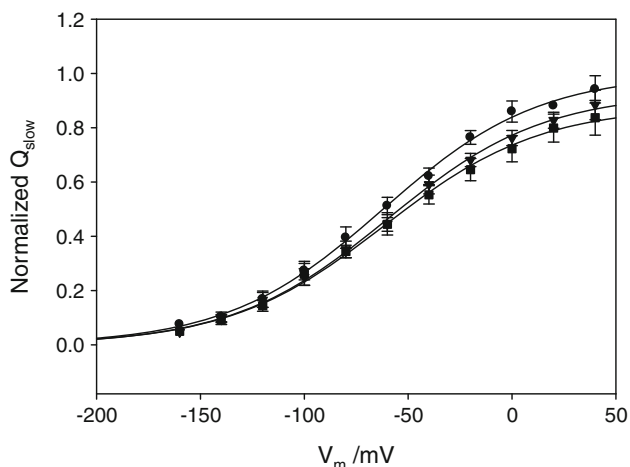
In the absence of ADP, Scheme 1 is reduced to



which includes the transitions eliciting the charge translocation that we measure (see the Appendix for more explanation). Therefore, we might expect that adding intracellular ADP should result in a reduction of the amount of charge.

Figure 6 shows how charge distributes ( $Q$ - $V$  curve) at three different values of  $V_h$ ,  $-20$  (circle),  $-60$  (triangle) and  $-100$  mV (squares), and in the absence of ADP. All data from each oocyte were normalized to the  $Q_{\text{tot}}$  obtained at  $V_h$  of  $-20$  mV. Then, data from five oocytes of at least two frogs were pooled together and fitted again to obtain the best-fit parameters  $Q_{\text{tot}}$ ,  $V_{\text{mid}}$  and  $z$ .  $Q_{\text{tot}}$  showed slight dependence on  $V_h$ .  $Q_{\text{tot}}$  was  $0.88 \pm 0.02$  at  $V_h$  of  $-60$  mV and  $0.82 \pm 0.03$  at  $V_h$  of  $-100$  mV.

Figure 7 shows the charge distribution from a similar set of experiments but in the presence of 5 mM intracellular ADP. Clearly, under the present experimental conditions, ADP has no effect on the amount of charge moved at different  $V_h$ , suggesting that ADP could not displace pumps toward the intracellular  $\text{Na}^+$ -binding/release transitions. This failure could be because there was not enough



**Fig. 7** The  $V_h$  dependence of the  $Q$ - $V$  curve of ouabain-sensitive transient currents in oocytes in internal solution 50 mM  $\text{Na}^+$ , 5 mM ATP, 5 mM ADP and 0  $\text{K}^+$  and externally perfused with 100 mM  $\text{Na}^+$  0  $\text{K}^+$ . Dots are data points (mean  $\pm$  SE) at  $V_h$  of  $-20$  mV (circles),  $-60$  mV (triangles) and  $-100$  mV (squares). Solid lines represent best fit to Eq. 1 with  $z$  constrained to 0.68. The  $z$  value was obtained by initially fitting the  $Q$ - $V$  curve at  $V_h$   $-20$  mV to Eq. 1 with  $z$  floating. The best-fit parameter values are  $Q_{\text{tot}} = 1.00 \pm 0.03$ ,  $V_{\text{mid}} = -62.2 \pm 4.8$  mV for  $V_h = -20$  mV;  $Q_{\text{tot}} = 0.93 \pm 0.03$ ,  $V_{\text{mid}} = -59.8 \pm 5.3$  mV for  $V_h = -60$  mV;  $Q_{\text{tot}} = 0.88 \pm 0.03$ ,  $V_{\text{mid}} = -61.5 \pm 6.7$  mV for  $V_h = -100$  mV. Data were from five oocytes from at least two different frogs

difference between intracellular and extracellular  $[\text{Na}^+]_i$  to drive the pumps backward.

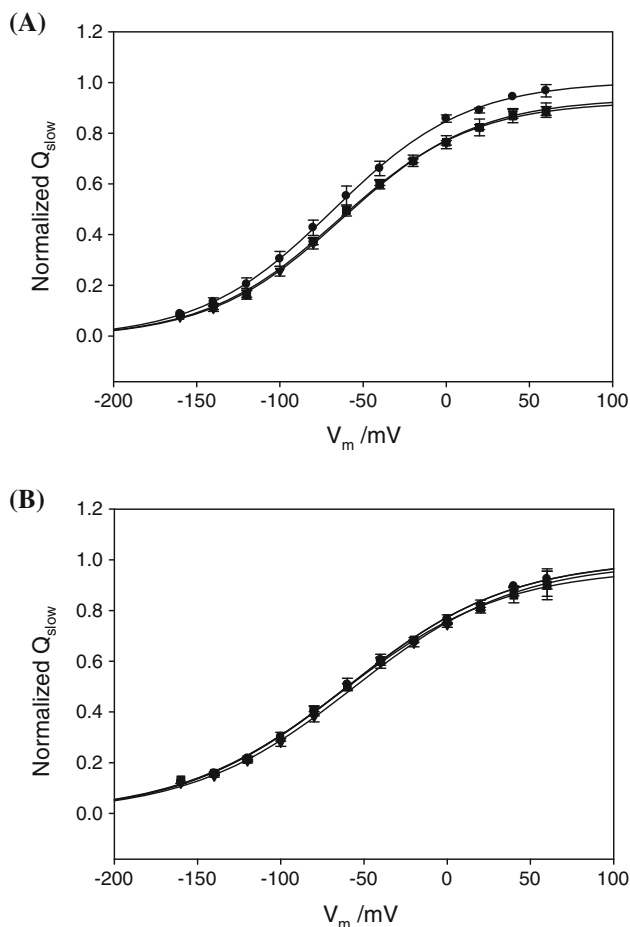
#### Effect of $V_h$ on the Magnitude of $Q_{\text{slow}}$ in 100 mM $\text{Na}_o^+$ and 10 mM $\text{Na}_i^+$ or 1 mM $\text{Na}_i^+$

In an attempt to increase the transmembrane difference in  $[\text{Na}^+]_i$ , we performed similar experiments in the presence of 5 mM ADP and 100 mM  $[\text{Na}^+]_o$  but decreased the intracellular  $[\text{Na}^+]_i$  from 50 to 10 or 1 mM. Previously, it was shown that  $K_m(0)$  for intracellular  $\text{Na}^+$  binding is 3.2 mM for endogenous *Xenopus* NKA (Holmgren and Rakowski 2006). Assuming that *Bufo* NKA  $\alpha_1\beta_1$  has a similar apparent affinity, then the changes of  $[\text{Na}^+]_i$  in our study would cover a large range of the dose-response curve.

Figure 8 shows that in the presence of 10 mM or 1 mM  $\text{Na}_i^+$   $Q_{\text{tot}}$  does not change with different holding potentials, suggesting that not even a 100-fold difference between intracellular and extracellular  $[\text{Na}^+]_i$  was enough to drive the pumps toward the inward-facing  $\text{Na}^+$ -binding sites.

#### Effect of $V_h$ on the Magnitude of $Q_{\text{slow}}$ at Low $[\text{Na}^+]_o$

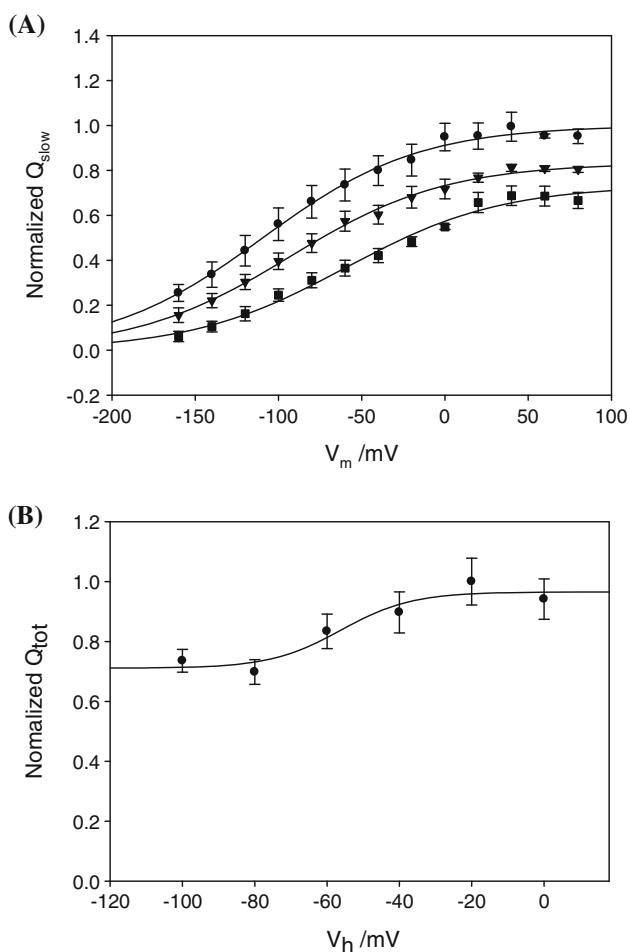
Figure 9a shows  $Q$ - $V$  curves in the presence of 25 mM  $\text{Na}_o^+$  obtained at different  $V_h$ . The other conditions in this set of data were 10 mM  $\text{Na}_i^+$ , 5 mM ADP and 5 mM ATP.



**Fig. 8** The  $V_h$  dependence of the  $Q$ - $V$  curve of ouabain-sensitive transient currents in 10 and 1 mM  $\text{Na}_i^+$ . Oocytes were incubated in internal solution (in mM: 5 ADP, 5 ATP and 0  $\text{K}^+$ ) and externally perfused with 100  $\text{Na}_o^+$  0  $\text{K}^+$ . **a** The voltage dependence of normalized charge ( $Q_{\text{slow}}$ ) in 10 mM at  $V_h$  of  $-20$  mV (circles),  $-60$  mV (triangles) and  $-100$  mV (squares). Dots are data points (mean  $\pm$  SE). Solid lines represent the best fit to Eq. 1 with  $z$  constrained to 0.63. The  $z$  value was obtained by initially fitting the  $Q$ - $V$  curve at  $V_h$   $-20$  mV to Eq. 1 with  $z$  floating. The best-fit parameters are  $Q_{\text{tot}} = 1.00 \pm 0.02$ ,  $V_{\text{mid}} = -67.9 \pm 3.7$  for  $V_h = -20$  mV;  $Q_{\text{tot}} = 0.95 \pm 0.01$ ,  $V_{\text{mid}} = -62.4 \pm 2.9$  for  $V_h = -60$  mV;  $Q_{\text{tot}} = 0.93 \pm 0.02$ ,  $V_{\text{mid}} = -64.7 \pm 3.7$  for  $V_h = -100$  mV. **b** The voltage dependence  $Q_{\text{slow}}$  in 1 mM. Symbols are the same as in **a**. The  $z$  value is 0.52, which was obtained by initially fitting the  $Q$ - $V$  curve at  $V_h$   $-20$  mV to Eq. 1. The best-fit parameter values are  $Q_{\text{tot}} = 1.00 \pm 0.02$ ,  $V_{\text{mid}} = -60.1 \pm 4.7$  for  $V_h = -20$  mV;  $Q_{\text{tot}} = 0.99 \pm 0.02$ ,  $V_{\text{mid}} = -56.5 \pm 5.0$  for  $V_h = -60$  mV;  $Q_{\text{tot}} = 0.97 \pm 0.02$ ,  $V_{\text{mid}} = -63.0 \pm 5.4$  for  $V_h = -100$  mV. Data were from five or six oocytes of at least two frogs

Interestingly, the more negative  $V_h$  was, the less  $Q_{\text{tot}}$  we could estimate, as if holding at negative potential effectively removed pumps from the states shown in Scheme 2. In fact,  $Q_{\text{tot}}$  appeared to follow a sigmoid distribution with  $V_h$  (Fig. 9b). If we assume that our system is losing pumps through a single voltage-dependent transition, then it would be rather strong voltage dependence with an apparent valence of 2.6.





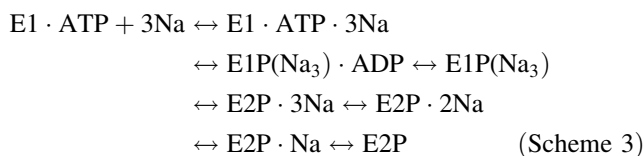
**Fig. 9** Effect of holding potential on ouabain-sensitive transient current in oocytes in internal solution 10 mM  $Na^+$ , 5 mM ADP, 5 mM ATP and 0  $K^+$  and externally perfused with 25 mM  $Na_o^+$ . **a** Voltage dependence of  $Q_{slow}$  at  $V_h$  of  $-20$  mV (circles),  $-60$  mV (triangles) and  $-100$  mV (squares). Dots are data points (mean  $\pm$  SE). Solid lines represent the best fit to Eq. 1 with  $z = 0.54$ . The  $z$  value was obtained by initially fitting the  $Q_{slow}$  vs.  $V_m$  curve at  $V_h = -20$  mV to Eq. 1 with  $z$  floating. The best-fit parameter values are  $Q_{tot} = 1.00 \pm 0.07$ ,  $V_{mid} = -109.0 \pm 15.8$  mV for  $V_h = -20$  mV;  $Q_{tot} = 0.83 \pm 0.05$ ,  $V_{mid} = -93.0 \pm 9.6$  mV for  $V_h = -60$  mV;  $Q_{tot} = 0.73 \pm 0.03$ ,  $V_{mid} = -59.8 \pm 8.6$  mV for  $V_h = -100$  mV. **b**  $V_h$  dependence of  $Q_{tot}$ . Values of  $Q_{tot}$  were obtained by fitting  $Q-V$  curves at six  $V_h$ , 0,  $-20$ ,  $-40$ ,  $-60$ ,  $-80$  and  $-100$  mV, to Eq. 1 with  $z = 0.54$  and then normalizing to the value obtained at  $V_h = -20$  mV. Dots are data points (mean  $\pm$  SE). Solid line is the best fit to the Boltzmann function. The best-fit  $z$  value is 2.6. Data were from four oocytes of two frogs

### Discussion

#### E1P(Na<sub>3</sub>)/E2P Occupancy Under Different Conditions

To interpret the results, we have to once again review the rationale behind transient currents. There exists a dynamic equilibrium between the intermediate states in electroneutral  $Na^+/Na^+$  exchange mode, which can be expressed in a

more detailed form of Scheme 3 (see the Appendix for more explanation) (Wuddel and Apell 1995).



The distribution of pumps between these states depends on the concentration of substrates and products and on the membrane potential. In this study, we varied the concentrations of [ADP],  $[Na^+]_i$  and  $[Na^+]_o$  and the holding potential,  $V_h$ .

The charge translocation that we measured derives from a subset of transitions of Scheme 3, which relate to extracellular  $Na^+$  binding/unbinding and occlusion/deocclusion— $E1P(Na_3) \leftrightarrow E2P \cdot 3Na \leftrightarrow E2P \cdot 2Na \leftrightarrow E2P \cdot Na \leftrightarrow E2P$ . Within these transitions, negative potentials will displace the system toward the  $E1P(3Na)$ , whereas positive potentials will shift the pumps toward the  $E2P$  state. If any experimental condition allows pumps to exit from this subset of charge translocation transitions, then we would expect a change in  $Q_{tot}$ . If the escape route is a voltage-dependent process, then the changes in  $Q_{tot}$  will be different depending on  $V_h$ , and these differences will provide information on the voltage-dependent steps that do not relate to the extracellular  $Na^+$ -binding/release transitions.

#### Effect of $V_h$ on the Magnitude of $Q_{slow}$ with 100 mM $Na_o^+$ , 50 mM $Na_i^+$ in the Presence and Absence of ADP

In the presence of ADP, it is expected that pumps in the  $E1P(Na_3)$  could, in principle, transit to  $E1P(Na_3) \cdot ADP$  and then deocclude and release  $3Na^+$  to the intracellular medium with the production of ATP. However, our results in Fig. 6 show that the presence of 5 mM ADP in the intracellular solution did not elicit significant change of  $Q_{tot}$  as  $V_h$  became more negative, similar to the results obtained in the absence of ADP (Fig. 6). Therefore, under both conditions, 50 mM  $Na_i^+$  and/or 5 mM ATP was able to keep pumps from exiting the subset of charge translocation transitions.

#### Effect of $V_h$ on the Magnitude of $Q_{slow}$ with 5 mM ADP and Varying $Na^+$ Present Intracellularly

According to Eq. 2, we were expecting that changing  $V_h$  would affect the  $Q(V)$  by a factor of  $[Na]_i/([Na]_i + K_m[V_h])$ , where  $K_m$  and  $V_h$  are related by

$$K_m(V_h) = K_m(0)\exp(-\lambda_i FV_h/RT) \quad (3)$$

where  $K_m(0)$  is the apparent affinity for intracellular  $Na^+$  and  $\lambda_i$  is the dielectric coefficient of intracellular  $Na^+$

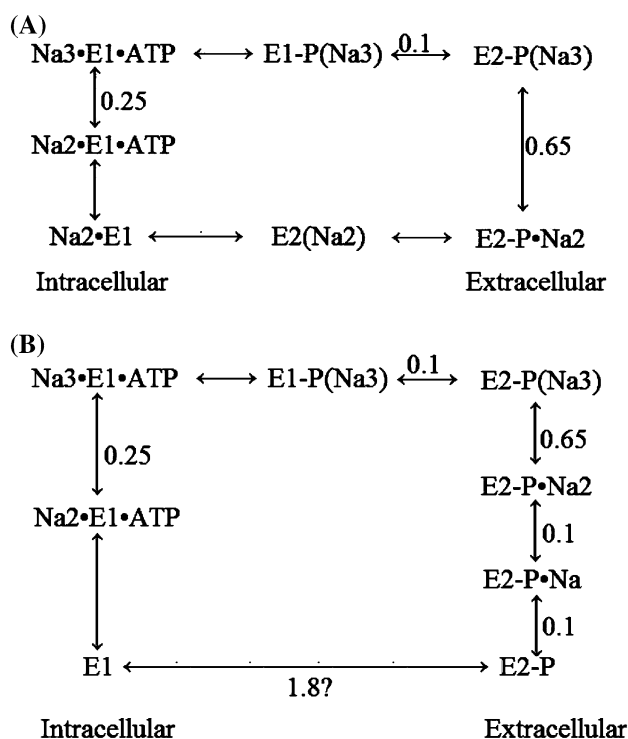
binding. We studied the effect of  $V_h$  on the pump-mediated charge translocation at three different  $[\text{Na}^+]_i$  (1, 10 and 50 mM). Of all conditions in our study, low  $[\text{Na}^+]_i$  was the condition that most favored pumps displacing toward the intracellular  $\text{Na}^+$ -binding steps. However,  $V_h$  has only minor changes in  $Q_{\text{tot}}$  in low  $[\text{Na}^+]_i$  (Figs. 7 and 8), indicating that we were not able to displace pumps toward the intracellular  $\text{Na}^+$ -binding steps. The most feasible explanation is that the magnitude of the forward rate is difficult to overcome by changes in occupancy of the E1(3Na)-P state with voltage, once the intracellular ATP is 5 mM.

#### Effect of $V_h$ on the Magnitude of $Q_{\text{slow}}$ in 25 mM $\text{Na}_o^+$

By reducing extracellular  $\text{Na}^+$  to 25 mM, we observed an unexpected result. At low  $\text{Na}_o^+$ ,  $Q_{\text{tot}}$  follows a sigmoidal shape with  $V_h$ , suggesting that negative  $V_h$  somehow remove pumps from the subset of charge translocation transitions. Furthermore, the route by which these pumps escape is a strong voltage-dependent transition.

What route is being followed by the pumps? In the absence of  $\text{K}^+$ , it has been reported since the 1960s (Garrahan and Glynn 1967) that pumps could operate in several modes. The most favorable is the traditional  $\text{Na}^+/\text{Na}^+$  electroneutral exchange, but under certain conditions the pump could transport either  $3\text{Na}^+/2\text{Na}^+$ , sometimes called  $\text{Na}^+$  ATPase activity (Fig. 10a), or  $3\text{Na}^+$  outwardly by an uncoupled ATPase activity (Fig. 10b) (Glynn and Karlish 1976; Heyse et al. 1994; Peluffo et al. 2000; Wuddel and Apell 1995).

Both transports are expected to be very slow processes. Therefore, experimental estimations of their reaction rate have been difficult. However, they have been estimated by numerical simulation utilizing other available values of the reaction cycle, and they are in the  $10^{-1} \text{ s}^{-1}$  range (Heyse et al. 1994). In principle, both of these noncanonical transport processes could provide escape routes for E2P. However, there are two reasons that suggest that uncoupled ATPase activity is the main source for escape. First, it has been previously documented that uncoupled ATPase activity is favored in low  $[\text{Na}^+]_o$ , contrary to the  $3\text{Na}^+/2\text{Na}^+$  ATPase activity, which is enhanced by high  $[\text{Na}^+]_o$  (Glynn and Karlish 1976; Heyse et al. 1994). In 50 mM  $\text{Na}_i^+$ , hyperpolarizing  $V_h$  causes a slight decrease in  $Q_{\text{tot}}$  in 100 mM  $\text{Na}_o^+$  (Fig. 7). This effect of  $V_h$  becomes more pronounced when  $[\text{Na}^+]_o$  decreases to 25 mM  $\text{Na}_o^+$  (Fig. 9). Second, our observation that the escape route is strongly voltage-dependent is consistent with previous theoretical estimations of the voltage dependence of uncoupled ATPase activity. According to the conservation principle, the sum of the dielectric coefficients of steps involved in uncoupled ATPase activity has to be 3 because three net charges are transported across the membrane in each cycle



**Fig. 10** Two possible operating modes of NKA other than electroneutral  $\text{Na}^+/\text{Na}^+$  exchange in the absence of  $\text{K}^+$ . **a** The  $3\text{Na}^+/2\text{Na}^+$  transport mode; **b** uncoupled ATPase activity with  $3\text{Na}^+$  efflux and no cation influx. Numbers beside lines indicate the dielectric coefficients that the particular partial reactions go through. Question mark indicates that the value was estimated by previous theoretical estimations (Heyse et al. 1994) and has not been verified by experiments

(Apell 2003). Of all partial steps in this cycle, only the dielectric coefficient of transition from E2-P to E1 was not acquired experimentally but estimated based on these known values and the conservation principle (Heyse et al. 1994).

The slow spontaneous transition from E2-P to E1 with a rate of about  $8 \times 10^{-1} \text{ s}^{-1}$  limits the turnover rate of uncoupled ATPase activity. The turnover rate of usual  $\text{Na}^+/\text{K}^+$  transportation is 20–100  $\text{ s}^{-1}$  (Boardman et al. 1972), which means that we could observe a steady-state current in the absence of  $\text{K}^+$  with about 1/100 magnitude compared to the  $\text{K}^+$ -activated pump current. The small magnitude of this current makes it very hard to record (Glynn and Karlish 1976). Future studies of the steady-state current mediated by the NKAs highly expressed in oocytes under low  $[\text{Na}^+]_o$  and  $\text{K}^+$ -free conditions will provide more information on the kinetics of uncoupled  $\text{Na}^+$  efflux.

#### Lack of an Effect of Varying [ADP] on Voltage Dependence of Slow Charge Movement

When  $Q$ - $V$  curves in 50 mM  $\text{Na}_i^+$ , 5 mM ADP (Fig. 5) were compared to the curves recorded at the same  $V_h$  in 50 mM  $\text{Na}_i^+$ , 0 ADP (Fig. 4), we failed to observe any difference at

$V_{mid}$ . For instance, at  $V_h$  of  $-20$  mV,  $V_{mid}$  is  $-62.2 \pm 4.8$  mV (mean and SE) in the presence of 5 mM ADP and  $-58.7 \pm 5.3$  mV in the absence of ADP. However, Peluffo (2004) recorded that increasing [ADP] caused the  $Q-V$  curve to shift to more positive  $V_m$ , which indicates that the presence of ADP in rat cardiac myocytes facilitated the backward reaction in  $Na^+$  translocation at half-cycle. A possible reason for this discrepancy is that the internal condition of 50 mM  $Na^+$ , 5 mM ADP and 5 mM ATP could not displace *Bufo marinus* NKA toward the intracellular  $Na^+$ -binding/release transitions, thus masking the subtle effect of [ADP] on the  $Na_o^+$ -dependent slow charge movement.

In summary, we studied the effect of  $V_h$  on  $Na_o^+$ -dependent slow charge movement mediated by *B. marinus* NKA expressed in *X. laevis* oocytes in different  $[Na^+]_i$ , [ADP] or  $[Na^+]_o$ . Lowering  $[Na^+]_i$  and increasing [ADP] were expected to influence the  $Na_o^+$ -dependent slow charge movement by displacing pumps toward the intracellular  $Na^+$ -binding steps. However, these variations did not show the effect of  $V_h$  on slow charge movement. The possible explanation is that these conditions were not strong enough to push *Bufo* NKA backward to release  $Na^+$  from  $E1P(Na_3) \cdot ADP$  to the internal medium when 5 mM ATP was present internally. However, unexpectedly, lowering  $[Na^+]_o$  brought out the inhibiting effect of hyperpolarizing  $V_h$  on slow charge movement. The strong  $V_h$  dependence of  $Q_{tot}$  in 25 mM  $Na_o^+$  suggests uncoupled  $Na^+$  efflux as a possible escape route of E2P in the absence of  $K^+$ .

**Acknowledgement** This article is dedicated to Dr. R. F. Rakowski. His earnest attitude to teaching and research and his open-minded attitude to people will always shine in our hearts. We sincerely thank Dr. M. Holmgren for his consistent and instructive help on data collection, data interpretation and writing after Dr. Rakowski passed away. We also thank Dr. J. Duerr and R. DiCaprio for their patient help with writing; Dr. J. D. Horisberger and Dr. O. Capendeguy for providing cDNAs of *Bufo* NKA  $\alpha_1$  and  $\beta_1$  and the reviewers for their constructive comments. Research was supported by NIH grant NS-022979 to R. F. R.

**Appendix: Equations and Reaction Schemes**

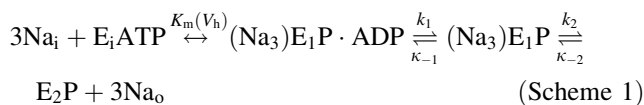
Reaction happening in electroneutral  $Na^+/Na^+$  exchange mode in the absence of ADP can be expressed as



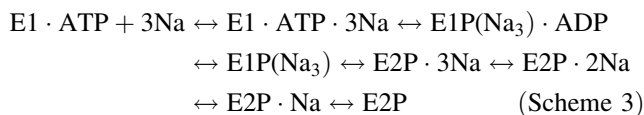
Under this condition, ideally there are no changes of  $Q_{tot}$  with  $V_h$ . The charge translocation can be expressed as a Boltzmann distribution:

$$\frac{Q(V) - Q_{min}}{Q_{tot}} = \frac{1}{1 + \exp\left[\frac{zF(V_{mid} - V)}{RT}\right]} \tag{1}$$

Reaction in the presence of ADP is extended and can be expressed as



Reaction in the presence of ADP can also be expressed as a detailed form:



Pumps can exit from the subset of charge translocation transition at any step of Scheme 3. Holmgren and Rakowski (2006) predicted pumps could exit to the left of  $E1P(Na_3)$  in Scheme 3, but the manipulation of  $[Na^+]_i$  and  $[ADP]_i$  in our present study failed to bring pumps to exit in this direction. Unexpectedly, in low  $[Na^+]_o$ , pumps might exit the subset of charge translocation transition through direct dephosphorylation of E2P to E1.

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